

## General Synthesis of Didehydroamino-acids and Peptides

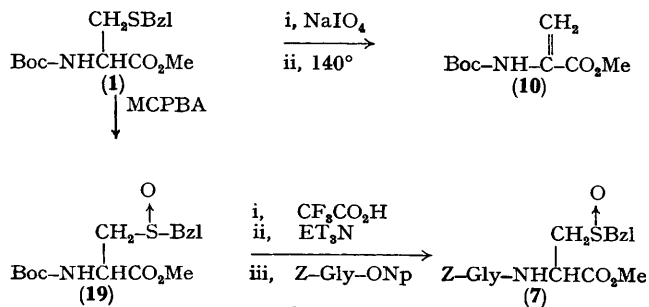
By DANIEL H. RICH,\* J. TAM, P. MATHIAPARANAM, J. A. GRANT, and C. MABUNI  
(School of Pharmacy, University of Wisconsin, Madison, Wisconsin 53706)

**Summary** Thermolysis of  $\beta$ -alkylsulphinyl derivatives of amino-acids and peptides gives  $\alpha\beta$ -didehydroamino-acids and peptides.

SEVERAL methods<sup>1,2</sup> for the synthesis of the  $\alpha\beta$ -didehydro-amino-acid unit have been developed for use in studies of the role of didehydropeptides in the biosynthesis<sup>3</sup> and mechanism of action<sup>4</sup> of microbial peptides. Additional synthetic methods are needed to prepare complex didehydropeptides containing hydroxy-amino-acids or peptide bonds susceptible to base-catalysed rearrangements. We report here the synthesis of didehydropeptides by thermolysis of  $\beta$ -alkylsulphinyl derivatives of amino-acids (Scheme).

Heating the sulfoxides (1)–(8) in refluxing xylene under nitrogen for 6–10 h gave the corresponding didehydro-compounds (10)–(17) which were isolated in good yield after chromatography on silica gel (Table). Elimination of the sulfoxide group was facilitated by replacement

of the amide hydrogen of the sulfoxide-containing amino-acid with a methyl group. The tertiary amide sulfoxide (9) was transformed in high yield into the didehydropeptide (18) within 48 h at room temperature in chloroform.



SCHEME

Bzl = benzyl

Ionization of the  $\alpha$ -proton in acylated secondary amino-acids has been reported to be suppressed by competing ionization of the more acidic amide N-H bond.<sup>5</sup>

The sulphoxides (1)–(9) were prepared by oxidation of the corresponding sulphides with sodium periodate in

which after treatment with  $\text{CF}_3\text{CO}_2\text{H}$  followed by neutralization with  $\text{Et}_3\text{N}$  and reaction with *p*-nitrophenyl *N*-benzyloxycarbonylglycinate gave the dipeptide sulphoxide (7) in 70% yield (Scheme).

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TABLE†

Sulphoxide	Didehydropeptide	Yield (%)
(1) Boc-Cys(OBzl)-OMe	(10) Boc- $\Delta$ -Ala-OMe	85
(2) Boc-But(3SOBzl)-OMe <sup>a,b</sup>	(11) Boc- $\Delta$ But-OMe <sup>b,c</sup>	81
(3) Boc-Val(3SOBzl)-OMe	(12) Boc- $\Delta$ Val-OMe	89
(4) Boc-Ala-Phe(3SOBzl)-OMe <sup>a</sup>	(13) Boc-Ala- $\Delta$ Phe-OMe <sup>c</sup>	78
(5) Z-Cys(OBzl)-Gly-OEt	(14) Z- $\Delta$ Ala-Gly-OEt	75
(6) Ac-Gly-Cys(OBzl)-OMe	(15) Ac-Gly- $\Delta$ Ala-OMe	80
(7) Z-Gly-Cys(OBzl)-OMe	(16) Z-Gly- $\Delta$ Ala-OMe	60
(8) Boc-MeAla-Leu-Phe(3SOBzl)-Gly-OMe <sup>a</sup>	(17) Boc-MeAla-Leu- $\Delta$ Phe-Gly-OEt <sup>c</sup>	65
(9) Boc-MePhe(3SOBzl)-Gly-MeAla-Leu-OMe <sup>a</sup>	(18) Boc-NMe $\Delta$ Phe-Gly-MeAla-Leu-OMe <sup>c</sup>	95

<sup>a</sup> The  $\beta$ -S-benzylamino-acid used to prepare the sulphoxides was a mixture of both *threo*- and *erythro*-diastereoisomers. <sup>b</sup> But =  $\beta$ -methylalanine. <sup>c</sup> The product obtained was a mixture of E and Z isomers.

aqueous methanol or with *m*-chloroperbenzoic acid in chloroform. Peptide sulphoxides can also be synthesized stepwise using preformed protected  $\beta$ -S-alkylsulphinyl derivatives of amino-acid and peptides. *E.g.*, oxidation of the  $\beta$ -S-benzylcysteinate (1) gave the sulphoxide (19)

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